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Activity Series
Post Lab
Experiment 7
Answers
Experiment 7

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~~How to Use the Activity
Series The Activity Series
Activity Series Lab
Activity Series and Single
Replacement~~

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~~Reactions.mp4 Activity Series of Metals \u0026 Elements - Chemistry~~

Activity Series Lab

Activity Series Virtual Lab Explanation Activity Series of Metals - Single Replacement Reactions Mr Pauller ~~Activity Series of Metals (Single Replacement): Observe \u0026 Record the Data~~

Activity Series \u0026

Pennies LabActivity

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Series Of Metals

Complete Lab Activity
Series Demonstration

Reactivity of Metals
with HCl - Qualitative
Lab ~~Keeping a~~

~~Laboratory Notebook~~

Doctor Reacts to Medical
TikTok Controversy

Reactivity of Metals with
water - Qualitative Lab

[4K] Displacement

Reaction of Metals - Zinc
in Copper (II) Sulfate -

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with explanation at micro
level E1 Lab Safety

Reactivity Series song

What is a Lab

Notebook?! Metal

Reactivity Series

Menomics Activity Series

of a Metal Lab 9.1

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Chemistry] TIMELAPSE

OF THE FUTURE: A

Journey to the End of

Time (4K) Pre-Lab

Activity Series of Metals

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(pg. 9) Displacement
Reactions - The
Reactivity Series Metal
Activity Series and
Oxides Activity Series of
Metals Laboratory v1
Reactivity Series of
Metals | Environmental |
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Activity Series Post Lab
Answers

The purpose of the lab
was to find which metal is
the most reactive and

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which metal is the least reactive. It was known before the experiment that the metals used in the experiment are placed in the activity series from most active to least active as follows: magnesium, aluminum, zinc, and copper. The hypotheses formed were that zinc nitrate would react with aluminum and magnesium; aluminum

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nitrate would react with
magnesium; copper
nitrate would react with
zinc, magnesium, and
aluminum; and ...

~~Activity Series Lab~~

~~Answers |~~

~~SchoolWorkHelper~~

Question: Laboratory 7

Relative Reactivities Of

Metals & The Activity

Series NAME: DATE:

: SECTION: POST-

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LAB REPORT Use The In-lab Observations To Complete The Laboratory Report. Turn In To Your Instructor When You Have Completed The Report.

PART A&B:

REACTION WITH

WATER 1. List The

Metals That Reacted

With Water In Order Of

Decreasing Reactivity

(most Reactive ...

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Post Lab
~~Solved: Laboratory 7
Answers
Relative Reactivities Of
Experiment 7
Metals & The ...~~

View Lab Report -
Activity series post lab
from CHEM 2038 at
University of Colorado,
Denver. Elizabeth Platt
Chemistry 2038
November 29, 2016
Exploring an activity
series post

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~~Activity series post lab—
Elizabeth Platt Chemistry
Answers
2038 ...~~

Create an activity series for the seven metals in this experiment by listing them from most reactive (at the top) to least reactive (at the bottom).

2. Answer the following questions based the activity series you just created. The word observed means what

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Activity Series

changes did you see
(such as color changes,
solid formation, etc.)
Write the equations ...

~~Metal Activity Series~~
~~Postlab Questions~~

Question:

EXPERIMENT 6:

RELATIVE

REACTIVITIES OF

METALS AND THE

ACTIVITY SERIES

Name: Instructor: Post-

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Lab Instructor When
You Have Completed
The Report.) PART A &
B: REACTION WITH

WATER Date:

Section/Group: Report

(Use The In-lab

Observations To

Complete The

Laboratory Report. Turn

In To Your Als That

Reacted With Water In

Order Of Decreasing

Reactivity (most ...

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Activity Series

Post Lab

~~Solved: EXPERIMENT~~
~~6: RELATIVE~~
~~REACTIVITIES OF~~
~~METALS AND ...~~

The activity series allows us to predict whether a metal displacement reaction will occur. ...

Answer. Yes. Magnesium is above copper on the reactivity series of metals. Therefore, it will replace the copper in the copper

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Activity Series

chloride, producing magnesium chloride and solid copper.

Experiment 7

~~Metal Activity Series~~
~~Chemistry | Socratic~~

The hypothesis has been proven after the experiment. The activity series were correctly found for the metals Magnesium, Zinc, Lead, and Copper and, the halogens Chlorine,

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Activity Series

Bromine, and Iodine by performing a series of reactions. This lab has almost a hundred percent accuracy since all the data found had matched the actual activity series.

~~An Activity Series~~ — Judy Chen

Answers to questions in complete sentences .

REVISED 12/2003

Activity Series Lab —

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Observations for Part 1

KNO_3 $\text{Mg}(\text{NO}_3)_2$

$\text{Zn}(\text{NO}_3)_2$ CuSO_4

AgNO_3 Distilled H_2O

Copper Iron Magnesium

Tin Zinc . REVISED

12/2003 Activity Series

Lab – Observations for

Part 2 Reaction with HCl

Copper

~~ACTIVITY SERIES LAB~~

~~–Auburn School District~~

After performing this lab,

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we were able to develop an activity series with Magnesium at the top (Being the most reactive) and Silver at the bottom (Being the least reactive), by comparing the reactivity of different metals in different metal and nonmetal solutions.

~~Shironaka Activity Series
Lab Report by Nick
Shironaka~~

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Activity Series

When an atom gains electrons, it is reduced. Metals higher on the activity series are more likely to react relative to those lower on the activity series. The activity series can be used to predict products of reactions, and to predict if a reaction will even occur. In this experiment, different metals were tested for

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Activity Series

their reactivity. It was recorded if a reaction occurred or not, so that an activity series could be created. Data & Results

~~Chemistry Lab Report
(The Activity Series)
Sarah Jackson~~

An activity series could also be created for the halogens. Describe a set of tests that you could perform in order to

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Post Lab
Answers
Experiment 7

accomplish this. 5. Post 1982 pennies have a zinc core with a thin copper outer shell. Using your activity series predict what would happen if a post 1982 penny were put into a solution of hydrochloric acid? 6.

~~Metal Activity Series~~
~~Postlab Questions~~

From this lab one can conclude that the activity

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Activity Series

series for the metals, from most active to least, is Magnesium (Mg), Zinc (Zn), Lead (Pb), Copper (Cu), Silver (Ag). The halogen activity series is Chlorine (Cl), Bromine (Br), Iodine (I). No sources of error. This experiment was qualitative, not quantitative.

~~An Activity Series Lab by~~

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~~av s~~ Prezi

AP Chemistry Lab 3 2

Activity Series of Metals
and Nonmetals 7

PROCEDURE Part 1.

Metals 1. Refer to Table 1 to see how the chemical solutions are arranged. 2. Thoroughly clean the spot-plate with soap and water. 3. Place about 3 drops of copper(II) nitrate solution in wells 2 through 4 in the first

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column. Put 3

~~Answers~~
~~AP Chemistry Lab 3-1~~
~~Experiment 7~~
~~Activity Series of Metals~~
~~and Nonmetals~~

The final activity rankings are Na and K Mg Zn Pb Cu Ag. From 5 – 8, Na and K Mg Zn. From 3 and 4, Zn Cu and Pb. Therefore, the order is Zn Pb Cu Ag. From 1 and 2, the relative activities are Pb Cu Ag. 8.

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K Mg (KCl Mg produce
no reaction) 7. Na Mg
(NaCl Mg produce no
reaction) 6. Mg Zn
(ZnCl₂ Mg produce a
reaction) 5.

~~REACTIVITY OF
METALS 15 Science
Curriculum~~

An Activity Series AP
Chemistry Laboratory
#20 Catalog No. AP5914
Publication No. 10536A I

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Activity Series

Introduction In this experiment, a series of metals and a series of nonmetal halogens are studied to find their relative reactivities. The reactivity of the metals is determined by combining the metals with a complemen-

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~~Weebly~~

View Notes - Lab 11

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Activity Series

(Metal Reactivity)

answers from CHEM 164
at Rutgers University.

Chemistry Lab (Demo)

Name_ Experiment 11

Metal Reactivity: (Single
Replacement Reactions)

Use the Activity Series

~~Lab 11 (Metal Reactivity)~~

~~answers—Chemistry~~

~~Lab (Demo ...~~

This introductory-level
activity can be used as a

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pre-lab to a unit on Mendelian genetics, and assumes that students are familiar with the terms genotype, phenotype, and allele. View »

Dealing with Data In this introductory lab, students collect data and then devise methods to organize and display the data to give it more meaning.

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~~Carolina Lab Sheets |~~

~~Carolina.com~~

In the Penny-Ante

Experiment 7
Equilibrium: A

Classroom

Activity—ChemTopic™

Lab Activity, pennies are

used as reactants and

products in a reversible

reaction to answer

questions about the

fundamental nature of

equilibrium and what

happens to the amount

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of reactants and products
when it is reached.

Experiment 7

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