

Chapter Test B Gases

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the sum of the partial pressures of the component gases. 2. Chapter Test B Gases Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1.

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Chemistry Chapter 14 The Behavior of Gases Test. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Aubr3y_Davis. Terms in this set (10) at constant volume and temperature the total pressure exerted by a mixture of gases is equal to the sum of partial pressures of the component gases.

Chemistry Chapter 14 The Behavior of Gases Test—Quizlet

(b) An English unit of pressure sometimes used in engineering is pounds per square inch (lb/in. 2), or psi: 1 atm = 14.7 lb/in. . If a pressure is reported as 91.5 psi, express the measurement in atmospheres. Answer: (a) 99.3 kPa, (b) 6.22 atm Practice Exercise 1) A gas at a pressure of 10.0 Pa exerts a force of ____ N on an area of A) 55

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Solids, Liquids, and Gases

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