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In a chemical reaction, one or more reactants change into one or more products. Figure 11.1a shows the ingredients for making leavened

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bread— flour, salt, yeast, and water.

Chemical reactions take place when the ingredients are mixed together and heated in the oven. Figure 11.1b shows the product—a loaf of bread.

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REACTIONS (pages 321–329) This

section explains how to write
equations describing chemical
reactions using appropriate symbols.
It also describes how to write
balanced chemical equations when

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given the names or formulas of the reactants and products in a chemical reaction. Writing Chemical Equations (pages 321–323) 1.

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(pages 321 Chapter 11 Chemical

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DESCRIBING CHEMICAL REACTIONS...

A chemical reaction occurs when one
or more _____ change into one

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Most complaints are temporary and self-limiting and do not need medication. Moreover, you must cite properties that change as a consequence of the reaction.

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(pages 321.

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Reactions Pages 321 329 ...

Chemical reactions can be
represented using equations.

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How can she conclude that the reaction started with three nitrogen atoms? There are three nitrogen atoms in the products. The law of conservation of mass states that the amount of substance should be the same before and after a reaction.

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Chapter 3 Chemical Reactions and
Reaction Stoichiometry Pp Remember
that these answers can be found in

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different kinds of factors. You may be looking for examples, ratios, weight, sizes, and so on.

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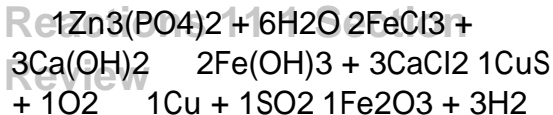
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On May 6, 1937, the huge air-ship Hindenburg was heading for its landing site in Lakehurst, New Jersey, after an uneventful trans-Atlantic crossing. Suddenly, to the horror of observers on the ground, the airship erupted into a reball. Within a short time,

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11

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 $1\text{Ag}_2\text{S} + 2\text{HNO}_3 + 3\text{Zn}(\text{OH})_2 + 2\text{H}_3\text{PO}_4$

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2Fe + 3H₂O

1. Propane gas reacts with oxygen to produce water vapor and carbon dioxide.

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