

## Modern Chemistry Review Stoichiometry Section 2 Answers

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1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of  $C_3H_4$ ? 2 mol O 2:1 mol H 2O c. What is the mole ratio ...

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Modern Chemistry Section Review Stoichiometry Answers Mix & match (both simple stoichiometry and limiting reagent problems) Units to use (select at least one): Grams Moles Particles (e.g. atoms/molecules/formula units) Chemical formulas or names: Formulas only Names only Mix & match formulas &

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\_\_\_\_\_ The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol

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### ~~Modern Chemistry Chapter 9 Review Answers Section 3~~

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